

ISOLATION OF APOHYOSCINE AND OF 6-HYDROXYATROPINE  
FROM *Physochlaina alaica*

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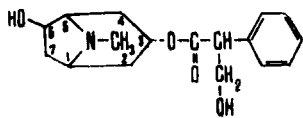
UDC 547.944

From the roots of *Physochlaina alaica* E. Korot. collected in the period of the withering of the epigeal part, in addition to compounds found previously [1], we have isolated two alkaloids.

The alkaloid (I),  $C_{17}H_{19}NO_3$ , has mp 79–80°C (ether),  $M^+$  285. Its IR spectrum shows absorption bands at 705 and 780  $cm^{-1}$  (monosubstituted benzene ring), 1700  $cm^{-1}$  (conjugated ester carbonyl group), and 1610  $cm^{-1}$  ( $>C=C<$ ). The UV spectrum of the alkaloid has a broad band at  $\lambda$  240–262 nm like the spectra of hyoscyamine [2]. In the NMR spectrum of (I) (in  $CDCl_3$ ,  $\delta$  scale, JNM 100/100 MHz, HMDS as internal standard) there is a singlet at 2.40 ppm (3 H,  $N-CH_3$ ), two one-proton doublets at 5.65 and 6.15 ppm ( $C=CH_2$ ), a singlet at 7.22 ppm (5 H, monosubstituted benzene ring), a triplet at 4.95 ppm ( $C_3H$ ) and a singlet at 3.17 ppm ( $C_6H$  and  $C_7H$ ). The molecular weight of (I) determined mass-spectrometrically differs from that of hyoscyamine by 18 m/e. The facts given above, and also the absence of a depression of the melting point of a mixture with an authentic sample shows that the base isolated is apohyoscyamine [3].

The alkaloid (II),  $C_{17}H_{23}NO_4$ , has mp 68–69°C (carbon tetrachloride),  $[\alpha]_D^{20}$  0° (c 1.9; methanol),  $M^+$  305. The IR spectrum of (II) has absorption bands at 710 and 745  $cm^{-1}$  (monosubstituted benzene ring), 1730  $cm^{-1}$  (ester carbonyl group) and 3320–3480  $cm^{-1}$  (broad) (hydroxy group). The UV spectrum has a broad band at  $\lambda$  250–270 nm similar to the spectra of the tropane alkaloids [2]. The NMR spectrum (in  $CF_3COOH$ ,  $\delta$  scale) has signals at 2.73 ppm (3 H, singlet,  $N-CH_3$ ), 4.82 ppm ( $C_3H$ , triplet), 1.45–2.35 ppm (6 H, protons at  $C_2$ ,  $C_4$ , and  $C_7$ ), 2.65–2.95 ppm (2 H, protons at  $C_1$  and  $C_5$ ), 4.50 ppm ( $C_6H$ ), 3.50–4.20 ppm (5 H,  $>CH-CH_2OH$ ,  $-OH$  at  $C_6$ ), and 6.99 ppm (5 H, singlet, monosubstituted benzene ring). The mass spectrum of (II) has peaks of ions with m/e 305 ( $M^+$ ), 261, 156, 140, 96, 95, 94, 83, 82, 81, which are characteristic for alkaloids of the tropane group [4].

The results of a comparison of the spectral characteristics of the alkaloid and of 6-hydroxyhyoscyamine, the production of a diacetyl derivative (III) (NMR spectrum: 1.93 ppm, 6 H, singlet,  $CCl_4$ ), and also the formation of tropane-3 $\alpha$ ,6 $\beta$ -diol and of d, l-tropic acid [5] on its hydrolysis showed that the new base has the structure of 6-hydroxyatropine.



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Order of the Red Banner of Labor Institute of the Chemistry of Plant Substances of the Academy of Sciences of the Uzbek SSR, and M. I. Kalinin Andizhan State Medical Institute. Translated from *Khimiya Prirodnikh Soedinenii*, No. 3, pp. 416–417, May–June, 1974. Original article submitted December 29, 1973.

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